

### Western Coatings Show 2017

A Novel Acrylic DTM with Next Generation Corrosion Resistance at 50g/L VOC

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# Outline

- Project Background
- Structure/Property Study
- Next Generation Corrosion Resistance
- Solventborne & Waterborne Benchmarks

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- Other Performance Properties
- Summary

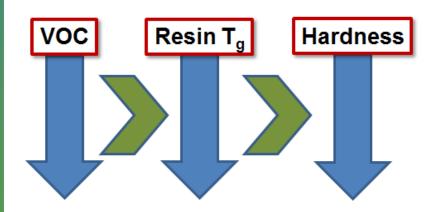
### **Project Background**

#### **Project Mandate**

Deliver best in class balanced, low VOC DTM resins

Lower VOC demands result in higher technical complexity in an effort to maintain full balance of properties

Interplay between adhesion and corrosion prompted in depth structure/property investigation



#### **Performance Tradeoffs**

Lower T<sub>g</sub> (or low VOC plasticizer) to lower VOC reduces hardness and block resistance

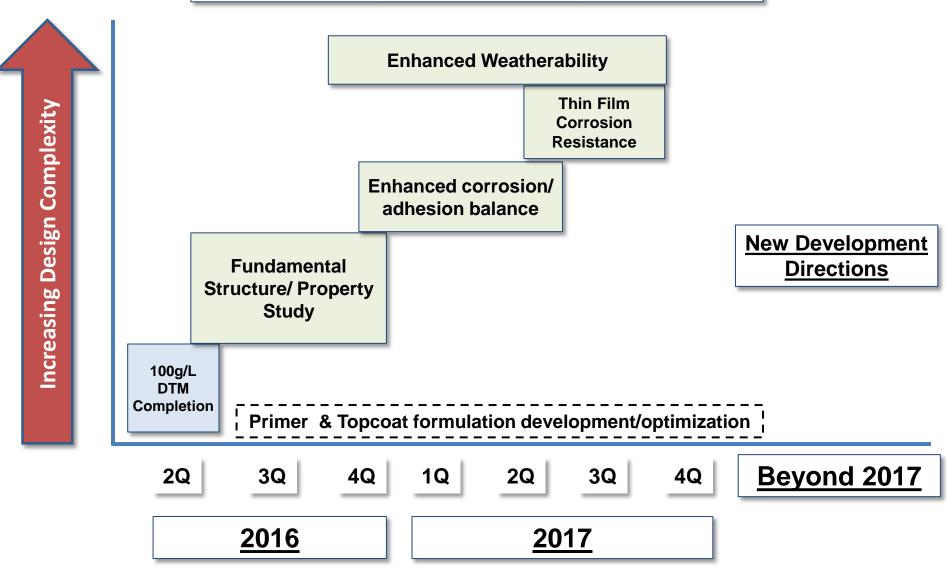
Some formulation mitigations available, but not ideal

Fluorosurfactants



### **Development Cycle**





## Literature Review

Barrier Properties	<ul> <li>S. Guruviah, JOCCA, 53, 1970, 660; P. Kresse, Pigment Resin Tech, 2(11), 1973, 21</li> <li>O<sub>2</sub> transport limiting factor in corrosion protection</li> </ul>
Impedance	<ul> <li>J. Mayne, <i>Corrosion</i>, 1976, pp15:24-15:37; Bacon, et al, <i>Ind Eng Chem</i>, 40(1), 1948, 161 <ul> <li>Films generally too permeable for barrier properties to be important</li> <li>Inhibition of galvanic cell via a high film resistance which impedes electrolyte transport most important factor</li> </ul> </li> </ul>
Adhesion	<ul> <li>W. Funke, H. Haagen, Ind Eng Chem Prod Res Dev, 17, 1978, 50;</li> <li>E. Parker, H. Gerhart, Ind Eng Chem, 59(8), 1967, 53</li> <li>– Loss of adhesion leads to onset of corrosion</li> </ul>

See appendix for further reading suggestions



#### **Corrosion Process - Steel** Requires $\mathbf{O}_2$ Water $O_2$ (CO<sub>2</sub>, or other reducible species) Electrolytic pathway $Fe^{2+} + 2OH^{-} \rightarrow Fe(OH)_{2}$ Water $4Fe(OH)_2 + O_2 \rightarrow Fe_2O_3 \bullet xH_2O$ $Fe \longrightarrow Fe^{2+} + 2e^{2}$ **Red Rust** Cathode Cathode Anode $O_2 + 2H_2O + 4e^- \rightarrow 4OH^ O_2 + 4H^+ + 4e^- \rightarrow 2H_2O$ e<sup>-</sup> e<sup>-</sup>

#### $Fe^{2+}+2H_2O \rightarrow Fe(OH)_2 + 2H^+$

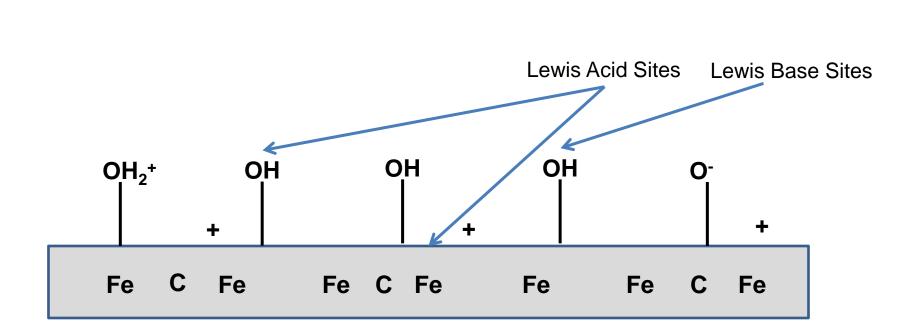
#### Possible mechanisms of corrosion prevention

Block water penetration `

**Barrier Properties** 

- O<sub>2</sub> transport inhibition
- Adhesion surface passivation, exclusion of water, etc
- Interference with electrolytic pathway coating resistance

## Adhesion to Steel



- Provided sufficient wetting is present, acid/base interactions, ionic interactions and van der Waals forces considered of primary importance<sup>1</sup>
- Isoelectric point of steel difficult to pinpoint, but likely around pH ~8-9
- As ammonia evaporates and pH drops, cationic sites arise allowing for electrostatic interactions

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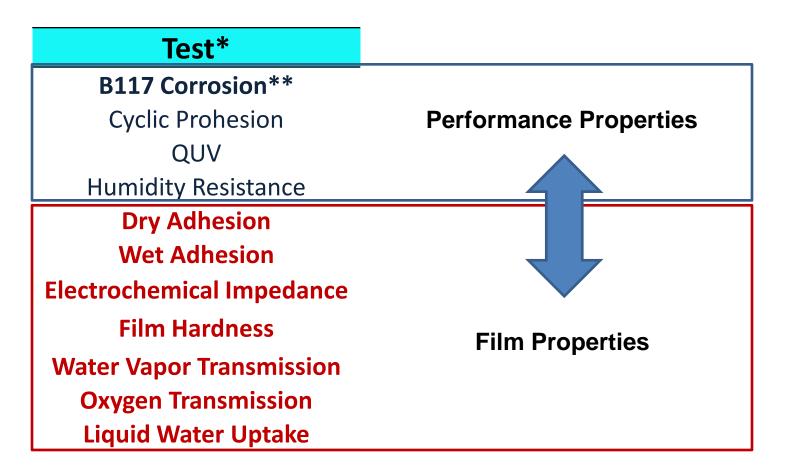
• Mechanical interlocking also significant in blasted substrates

<sup>1</sup>Fowkes, F.M., J Polym Sci J Polym Chem Ed, 1984, 22, 547

# **Study Description**

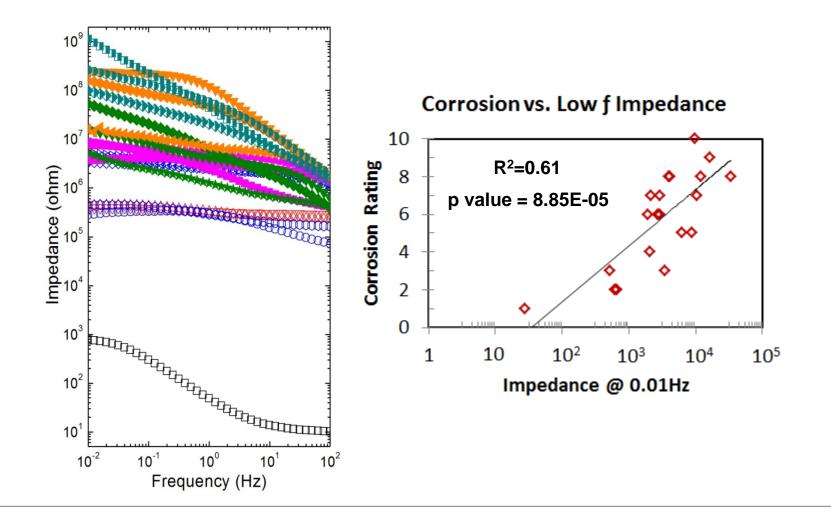
- 21 internal and external styrenated acrylic systems
- Wide variety of performance capabilities, Tg's, MFFTs, etc
- Formulated into single clear formulation adjusted coalescent level for MFFT
- Evaluated in a number of performance tests to develop a film property/corrosion model

## Study Design





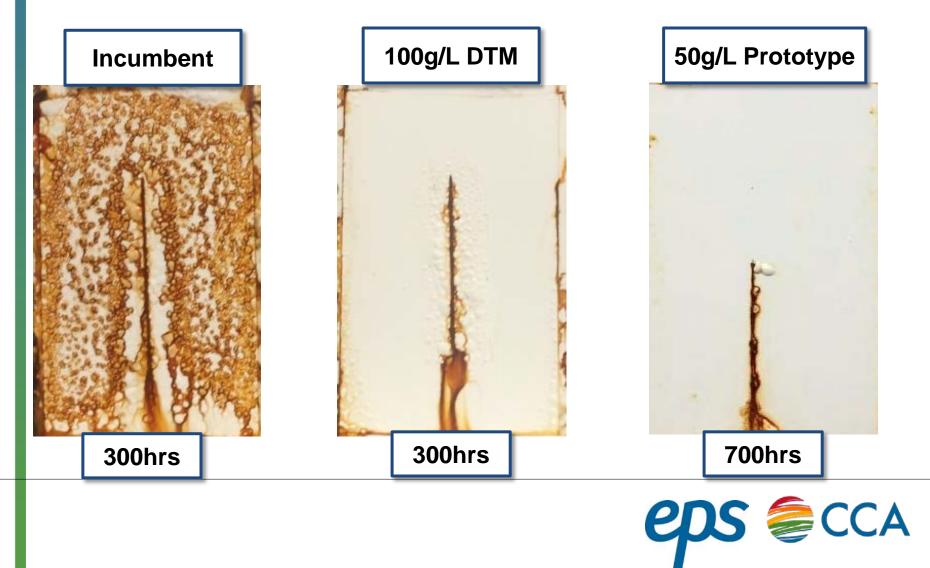
## **Impedance Studies**

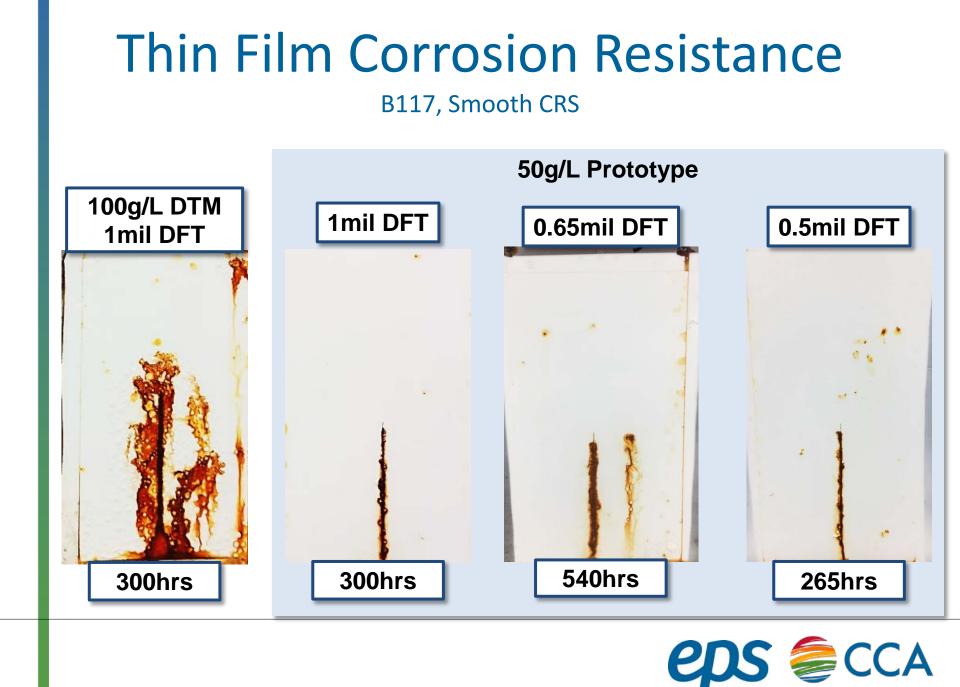


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## **Corrosion Progression**

#### B117, 2mil DFT, Smooth CRS





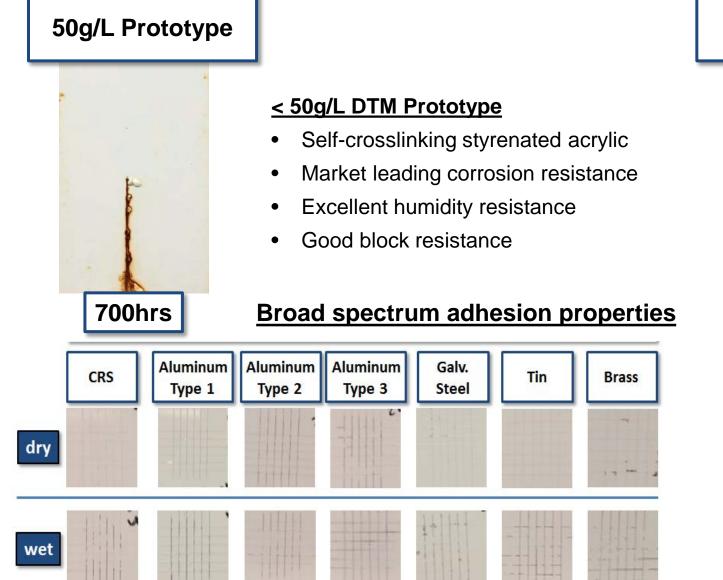
### **Next Generation Development**

#### B117, 2mil DFT, Smooth CRS

Competitor

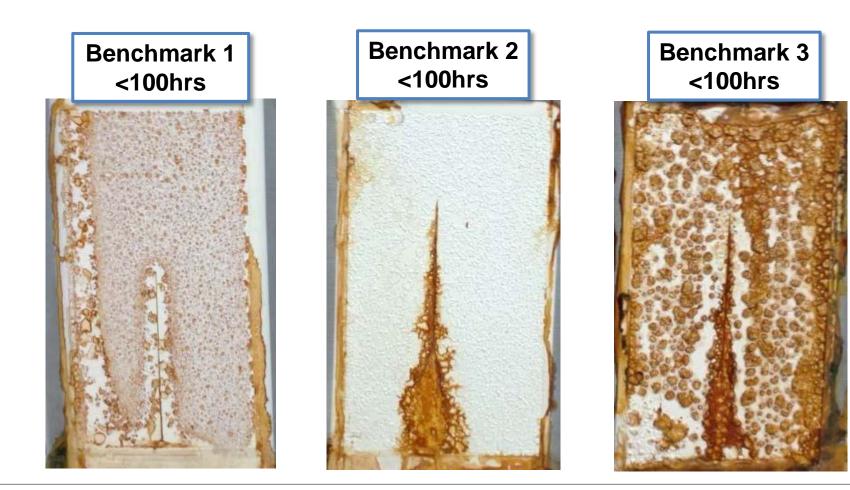
50g/L DTM

350hrs



### **Commercial Benchmarks**

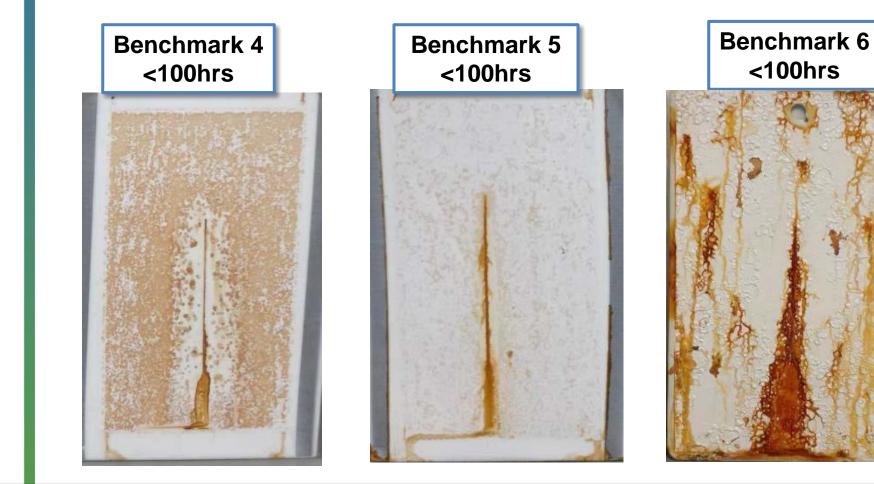
#### B117, 2mil DFT, Flat CRS



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### **Commercial Benchmarks**

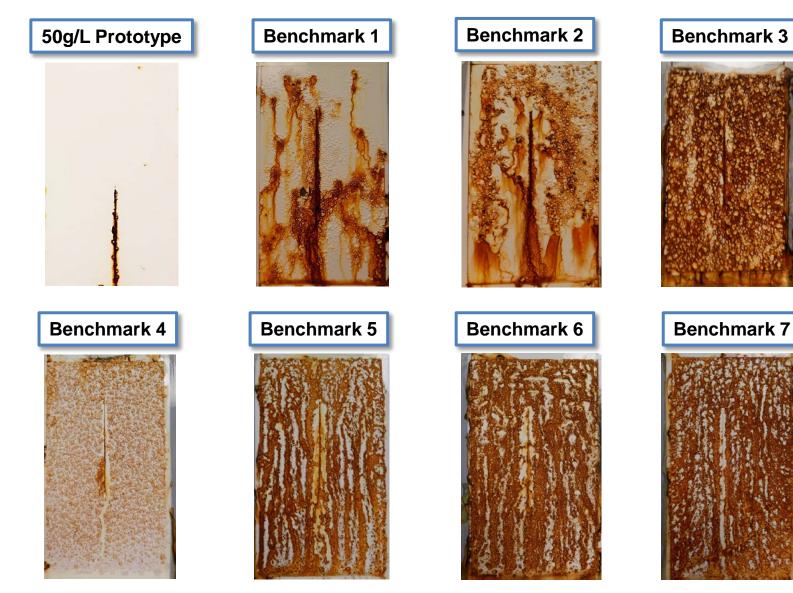
#### B117, 2mil DFT, Flat CRS



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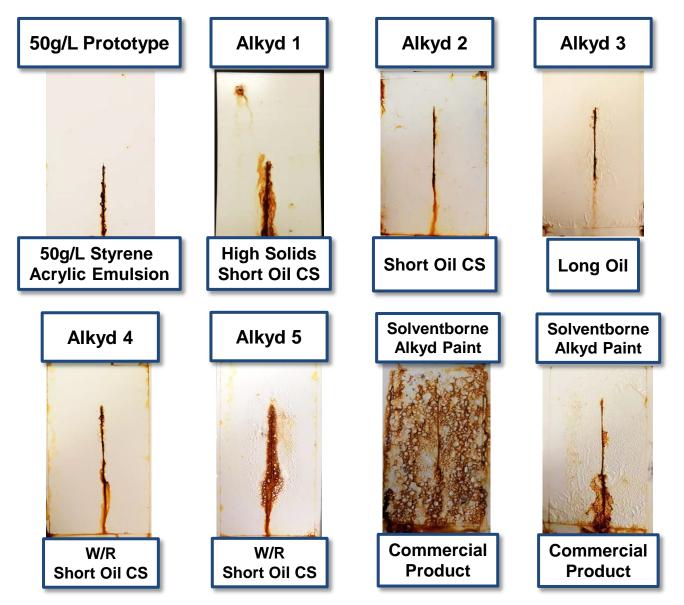
### Waterborne DTM Benchmarks

#### B117, 1mil DFT, Flat CRS, 300hrs

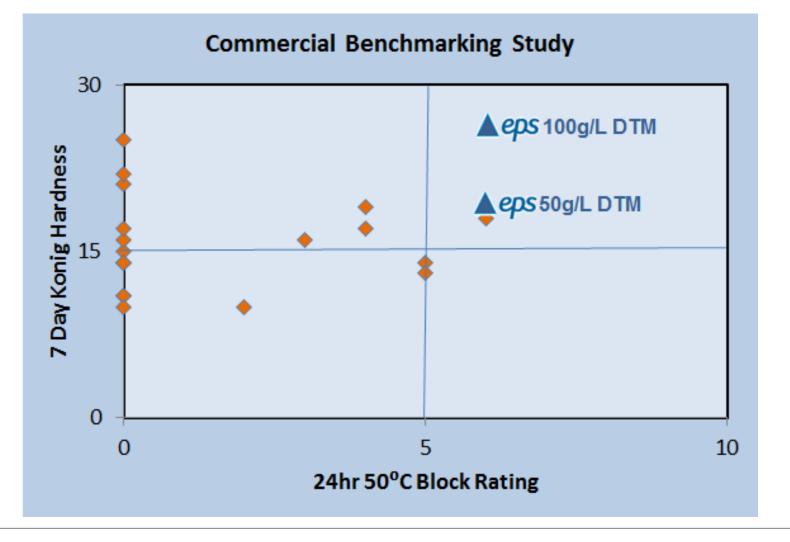


### **Benchmarking Solventborne Alkyds**

B117, CRS, 1mil DFT, 300hrs



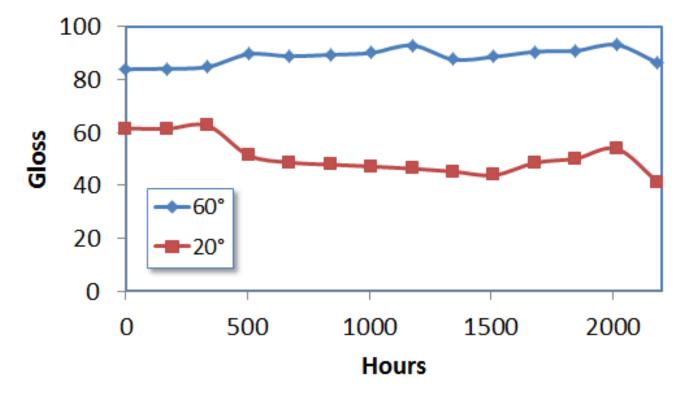
## Hardness/Block Survey



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### **Extended Weatherability**

#### **QUVA Performance**



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## Summary

- Protective action of coatings on steel is a complex process with multiple modes of action
- Experimental observations and more recent studies point towards impedance as primary predictor of corrosion protection in B117 salt fog
- Learnings leveraged toward development of 50g/L DTM with previously unachieved corrosion resistance
- Performance is comparable to and/or exceeding that of solventborne and water reducible alkyds at thin films
- The new low VOC DTM delivers a balance of properties including corrosion resistance, humidity resistance, multi-substrate adhesion, block resistance, and weatherability
- Performance optimization continues through formulation development and exploration of functional additives



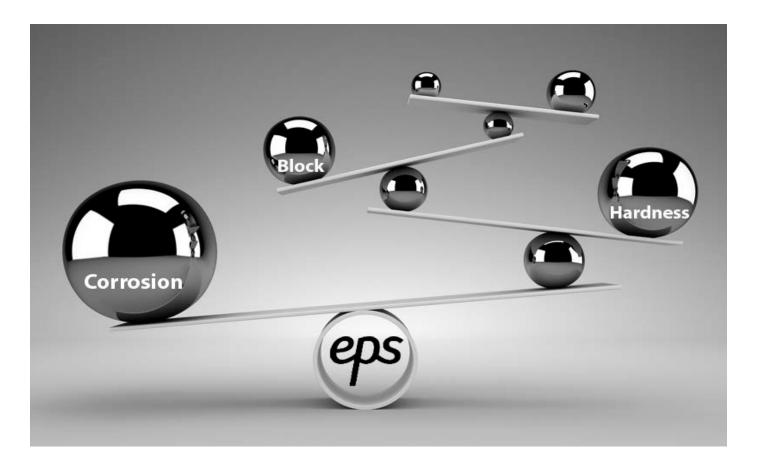
## Acknowledgements

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- Howard Killilea
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- Morgan DeWall
- MJ Hibben

### **Questions?**





## **Further Reading**

- R.A. Dickie, F.L. Floyd, "Polymeric Materials for Corrosion Control," ACS Symposium, 1986
- D. Greenfield, D. Scantlebury, "The Protective Action of Organic Coatings on Steel: A Review," *JCSE*, 3(5), 2000
- E. van Westing, "Determination of coating performance with impedance measurements," TNO Centre for Ctgs Res, 1992
- M. A. Butt, et al, "Theory of Adhesion and its Practical Implications," *J Faculty of Eng & Tech*, 2007-2008, 21-45

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- G. Bierwagen, et al, Prog in Org Ctgs, 46(2), 2003, 149
- F. Floyd, et al, *Prog in Org Ctgs*, 66(1), 2009, 8
- M. O'Donoghue, et al, *Coatings & Linings*, 2003, 36
- S. Shreeptahi, et al, J Coat Tech & Res, 8(2), 2011, 191
- C. Moreno, et al, Int J. Electrochem Sci, 7, 2012, 8444